

Surfacing – Technologies and Layer Properties (a Review)

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Abstract—The overview includes the basic methods for obtaining surface layers: arc surfacing - submerged arc surfacing (SAS), tungsten inert gas arc surfacing (TIGS), metal gas surfacing (MIG/MAG), plasma surfacing (PS); electron beam surfacing (EBS); laser surfacing (LS); electroless cladding (ELC); friction stir surfacing (FSS). Layers are applied on carbon steels, low alloyed steels, highly alloyed steels, aluminium alloys, titanium alloys or ductile cast iron substrates. The data on the additive materials used are provided and data on the hardness and wear resistance of the layers is given.

Keywords— surface layers; methods; hardness, wear.

I. INTRODUCTION

The application of layers and the modification of the surface of the parts are widely used in various areas of the industry. This review examines various technological processes that allow for a substantial change in surface properties relative to those of the substrate. Emphasis is placed on the resulting hardness and surface wear. The wear resistance data of the layer is generally compared with that of the substrate metal in the application of layers of homogeneous composition. The wear resistance of the layers in which there is a dispersed phase is compared to layers without dispersive phase. The variants, which demonstrates the best results are shown. The hardness data are given in the way they are published by the authors concerned, indicating the maximum values achieved.

II. TECHNOLOGICAL PROCESSES

The surfacing technological processes are in most cases close to welding processes. However, there are also technological variants that differ significantly from welding and are not applicable to joint parts. Thus, the thickest layers are obtained by electroslag surfacing (ESS) [1]. In this case, as with submerged arc surfacing (SAS) [5], [6], [7], [19], the layer characteristics are obtained depending on the combination of the components contained in the flux and the additive wire. Typically, the additional wires are solid, but core wires [6], [19] is used also. The components added in the flux which leads to increasing of the hardening phases [2] results in similar effects on hardness and wear resistance. The core wire are widely used in another electric arc processes - TIG [11],

[12] and [13] and MIG/MAG [3], [10], [15], [16], [17] 20]. The solid wires are applicable in the MIG / MAG process [4], [8], [9], [10], [18] too. Additional impacts such as forced gas cooling [4], longitudinal electromagnetic field [10] and post-surfacing electron beam treatment [3] are also used to improve the characteristics of the applied layer. The heat output of the welding arc is also used for remelting of the pre-deposited layer [14]. The plasma arc can also be used to remelt a deposited layer [21] or apply a layer using the powder injected in the plasma arc [22], [23], [24]. In both cases, the surface of the substrate melts, a welding pool is formed, and during on its crystallization the additive materials participate in the formation of the layer and thus determine its properties. When the aim is to obtain intermetallic compounds with the metal of the substrate in the coating, this can be achieved by adding the second metal in the form of a powder [23]. High cooling rates using heat sources with high energy concentration such as electron beam surfacing (EBS) [25], [26], [27], [28] and [29] or laser surfacing (LS) [30] - [39] creates conditions for modifying the surface layer even without the use of filler materials [25] or after a surfacing process has been performed [3], [26]. The electron beam process is also used for sintering [27] of powder mixture to form layers. Laser surfacing is realized in three ways. The first one involves the introduction of powder material into the heated zone [30], [31], [32], [33], the second assumes the remelting of a precoat [35], [36], [37] [39] and the third uses a gaseous medium to introduce the necessary components [34]. Another possibility to produce the layer is electroless cladding (ELC) from a solution and subsequent thermal treating [40], [41], [42]. The specific in this case is that no heating is required for the deposition of the bed, and the thermal treatment is at relatively low temperatures (2900C). Both in this case and in the friction stir surfacing (FSC) [43], [44], [45], [46], [47], the melting temperatures of the substrate are not reached. Therefore, diffusion processes have a significant effect on the final result. Powders may also be used during friction stir technology [44] and [45].

III. RESEARCH METHODS

The methods used to study layers properties can be

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divided into the following main groups. The microstructure is examined by optical (OM) [1], [9], [17], [21], [23], [27], [30], [31], [36], [42], [43] and electron microscopy (SEM) [1], [10], [11], [14], [21], [22], [23], [33], [39], [46], [47]. The phase composition is determined by the X-ray diffractometer [18], [23], [27], [31], [33], [36], [39], [45], the transmission diffraction electron microscopy [18] and energy-dispersive X-ray spectroscopy [14], [22], [30], [33], [39]. The X-ray fluorescence method [7] and the atomic emission spectrometer [7] were used to determine the chemical composition. Image analysis is used to determine the relative share of individual components [1].

IV. SUBSTRATE MATERIALS

In the publications reviewed, the materials used as substrates can be divided into the following main groups: carbon steels [3], [6], [7], [12], [13], [15], [16], [17], [18], [21], [25], [26], [29], [36] and [37]; alloyed steels [1], [4], [8], [9], [20], [22], [24], [30], [41] and [42]; highly alloyed steels [5] and [32]; aluminium and aluminium alloys [11], [28], [39], [43], [44] and [45]; titanium and titanium alloys [23], [31], [34] and [35]; ductile cast iron [33] and [40]. The SAS, MIG / MAG, TIG, PST, EBS and LS methods were used to apply the layers to carbon steels. When the substrate is made from alloy steel ESA, MIG / MAG, PST, LS and ELC were used. With a high alloy steel substrate, the SAS and LS methods are listed. The deposition of layers on ductile cast iron is represented by the LS and ELS methods. The TIG, EBS, LS and FSS methods have been used for layers deposited on aluminium or aluminium alloys. To deposit layers over titanium and titanium alloys two methods are presented - PST and LS.

V. FILLER MATERIALS AND LAYER PROPERTIES

Highest thickness layers are obtained using electroslag surfacing. The process for depositing high-chromium cast iron on low-alloy steel is described [1]. The chemical composition of the applied layer is shown in Table 1 [1]. The content of vanadium is varied. Rockwell hardness and wear resistant of the applied layer are measured and data (presented as graphics in [1]) is shown into the same table. The results show that hardness and wear resistance have the highest values for vanadium content of 1.5 wt%.

hardness of the specimen surface is also shown in Table 2. In general, it correlates with wear resistance. Experiments with a powder mixture containing (wt%): $Al_2O_3 = 21-46.23$; $F = 18-27$; $Na_2O = 8-15$; $K_2O = 0.4-6$; $CaO = 0.7-2.3$; $SiO_2 = 0.5-2.48$; $Fe_2O_3 = 2.1-3.27$; $C = 12.5-30.2$; $MnO = 0.07-0.9$; $MgO = 0.06-0.9$; $S = 0.09-0.19$; $P = 0.1-0.18$ are carried out [6]. A regression analysis of the obtained results was made and equations for the hardness and wear resistance of the type:

$$y = A_0 + A_1C + A_2Si + A_3Mn + A_4Cr + A_5Ni + A_6Mo + A_7Co + A_8Al + A_9Cu + A_{10}H \quad (1)$$

The values of the coefficients are given in Table 3.

The method presented in [19] uses two electrode wires for submerged arc surfacing- the first wire is equivalent to steel 30XGCA and second to aluminium alloy AlMg3 [19]. As mentioned, the second wire behind the main arc has a radically different chemical composition. The reached hardness of the layer varies from 28 to 55HRC depend on the conditions of the process - both filler wires ratios and the magnitude of the current.

In MIG / MAG overlaying [3], [9], [14], [15] and [18], core wires are commonly used. Table 4 shows the chemical composition. In [18] the Vickers hardness was also measured. Maximum measured value is 11 [GPa]. The main alloying elements are C-Mn-Si-Cr-Ni. In [14] layers are deposited on an aluminium substrate. Micro particles of SiC are used to increase wear resistance. Particles are coated with nickel using electroless method in salt ($NiSO_4$ and $NiCl_2$) solutions. In addition, the influence of flux (based on cryolite and three salts - sodium chloride, potassium chloride and magnesium chloride) deposited on the nickel coating was investigated. The results are compared with those obtained by the TIG method [11], where micro and nano SiC particles were also used. It has been found that wear resistance is highest when using nanoparticles coated with flux.

TABLE I

Layer chemical composition, wt% (Fe-bal.)					HRC	Volume lost, [mm ³]
C	Mn	Si	Cr	V		
3.62	2.31	0.86	19.48		58.5	220
3.59	2.25	0.68	20.19	0.83	59	160
3.58	2.06	0.68	19.40	1.50	61	80
3.56	1.91	0.64	19.57	2.32	60	180

The filler materials used with submerged arc surfacing are shown in Table 2 [5], [7] and [20]. In practice, in all cases, the major alloying elements are C-Mn-Si-Cr-Ni-Mo-V. To increase amount of carbides B and Nb are added. Small amounts of cobalt are also registered in [7] and [20]. Filler materials are solid or core wires. The resulting

TABLE II

Chemical composition (Fe-bal.) (wt%)									HV ₃₀	HRC	Wear rate		Remark
C	Mn	Si	Cr	Ni	Mo	V	B	Nb			mm ³ /(N.m)	mg/min	
0.12	1.0	0.6	12	2.92					400		0.052		filler
0.1	1.0	0.6	12	2.5	0.8	0.15		0.15	550		0.032		
0.25	1.0	0.6	9	0.25	2.0				600		0.0011		
0.3	1.0	0.6	12.2		0.75	0.15			675		0.0014		
0.19	0.52	0.74	2.79	0.17	0.26	0.14	0.001			44		0.03	deposit
0.22	0.62	0.35	2.78	0.09	0.25	0.02	0.001			36		0.04	
0.27	0.68	0.49	4.61	0.36	0.42	0.01	0.001			48		0.04	
0.43	0.84	0.37	7.04	0.42	0.49	0.03	0.001			56		0.02	
0.16	0.32	0.62	0.25		0.11	0.19	0.003			37		0.15	
0.10	0.45	0.15	0.23	0.08	0.10	0.11	0.003			28		0.10	
0.23	0.66	0.25	0.95	0.34	0.35	0.29	0.003			44		0.12	

TABLE III

A _i	Hardness		Wear resistance	
	with H	with no H	with H	with no H
	.10 ³			
A ₀	214.819	535.343	0.260	2.005
A ₁	-163.253	168.120	-0.695	1.110
A ₂	-307.499	-276.437	-0.973	-0.804
A ₃	-353.293	-890.442	-0.597	-3.522
A ₄	-21.239	6.037	-0.108	0.040
A ₅	33.871	108.851	0.091	0.500
A ₆	858.517	445.851	2.707	0.460
A ₇	-280.917	-433.688	-0.229	-1.061
A ₈		-1014.594		-5.525
A ₉	321.784	477.567	2.283	3.132
A ₁₀	13.830		0.075	
Appr. error, %	0.01	0.01	10.74	0.32

The TIG method is used to apply layers containing nanoscale particles of Al₂O₃ and TiCN. Nano powder is fed directly into the low-temperature zone of the welding pool. The method of direct nanoscale particle delivery is compared with the method of melting a pre-coated layer. The wear is compared to a sample without a nanomodifier. The results obtained show an increase in the wear resistance of the layers with modifier (up to 30%) and that direct delivery of the powder in the welding pool gives a better result. The achieved microhardness is up to 350 HV_{0.1}. The use of nanoscale particles of TiN, doped with chromium, as a dispersed phase, has been investigated in [13]. Overlayed layers with and without a nano modifier are compared. It has been found that the use of TiN + Cr increases up to 2 times the wear resistance. The hardness reached 282 HV.

TABLE IV

Chemical composition (Fe-bal.) (wt%)										Remark
C	Mn	Si	Cr	Ni	Mo	V	W	Nb	Cu	
1.3	0.9	1.1	7.0				1.4	8.5		filler
0.07	7.03	0.78	19	8.94	0.15				0.1	deposit
0.09	5.32	0.83	14.68	6.62						deposit
1.43	0.81	0.55	5.22	0.1	1.30			5.85		
4.46	0.76	0.77	23.54	0.11						
1.4			7.0			1.0	1.2	8.0		

When plasma surfacing process is used, the filler material is in the form of a powder. It can be fed through the plasma gun or to be pre-applied onto the pad in the form of a paste. Thus, for example, in the use of a powder containing (wt%) C-0.9; Si-2 ÷ 3; Cr-8 ÷ 10; B-2.8 ÷ 3.2 and Fe-bal. a hardness of the 60.6 HRC layer was obtained at the hardness of the 41 HRC matrix [22]. The wear of the layer compared to the substrate in this case is 30% less. In [24] results were obtained using a nickel based composition containing (wt%) of F3-2.5; C-0.3; Cr-7; B-2.2 and Ni-bal. The hardness reached in this case is 754 HV. In [21] we have shown a hardness of 400 ÷ 1800 HK using NiCrBSiC filler material. In [23] unalloyed aluminium powder is used and a layer is applied to titanium. The particle size is 75 to 147 [nm]. A TiAl-based intermetallic layer is obtained on the surface. Electron beam processes are mainly used to improve the properties of the applied layer. Using electron beam scan of the applied layer, the hardness increased 1.7 times and the wear resistance was 70 times higher comparing to substrate [25] and [26]. The main reason for this is the change of structure (a cell structure of 20 ÷ 100µm is obtained) on the surface of the layer due to the high cooling rates. The additive material contained (wt%): C-1.3; Mn -0.9; Si-0.3; Cr-7.0; W-1.4; Nb-8.5 and Fe-bal. In [27] the possibility of sintering of pre-compressed powdered samples containing TiC and Ti was investigated and the titanium content was (vol.%) - 50 and 60. In parallel, mixtures of TiC and 20 ÷ 50 % nickel based self-

fluxing alloy was examined. These mixtures could be used as a component of a paste applied to the substrate. The ability to obtain a layer of TiCuN on an aluminium substrate is shown in [28] through process modelling and experimental

verification. The layer was applied by plasma surfacing and subsequent electron beam treatment. Electron beam technology for deposition of layers by surface remelting is described in [29]. The samples was scanned with an electron beam to make specific roughness. On prepared rough surface a paste containing TiN or TiCN is applied. Subsequent remelting of the surface layer was performed. The wear resistance of the layer was 36% higher than that of the base reference sample without coating.

The materials used for deposition of layers by laser technology are given in table. 5. Various techniques are used to obtain the necessary hardening phases. Thus, for example, to obtain titanium nitride into the surface layer in the Ti_6Al_4V substrate, a process is carried out in a pure nitrogen medium [34]. The resulting hardness is in the range $1000 \div 2300 HV_{0.2}$. A mixture of Ti and TiC is introduced directly into the molten metal applying a layer over ductile cast iron [33]. Depending on the process parameters different concentrations of Ti are obtained in the melt layer from 3.0 to 5.9 and TiC - $4.8 \div 14.8 vol\%$. The reached hardness of the layer is $800 HV_{0.2}$. The wear resistance of the layer is about 10 times higher than that of the substrate. On the same type of substrate (Ti_6Al_4V) using a mixture of Ti_6Al_4V and B_4C is applied a layer by direct feeding of the powder material to the melting zone [31]. The hardness achieved in this case is 480 HV. On pure titanium a layer is applied using a graphite powder mixed with polyvinyl alcohol, dried and applied as a paste [35]. Surface is remelted. The achieved hardness of the layer surface is $300 HV_{0.3}$. Similarly, a graphite powder blended with Ni, C, V, and Ti is applied in the indicated proportions onto a medium carbon substrate [36] and [37]. The hardness of the resulting layer is $800 HV_{0.3}$. It has been found that the hardness can be increased to $970 HV_{0.3}$ by increasing the laser beam scanning speed. Powder-based paste [39] is applied to an aluminium substrate and scanned with a laser beam. The hardness of the layer is 500 HV. When the layer consists mainly of Al_3Ti / TiC its wear resistance increases to 10 times compare to the pad.

When layers are depositing on highly alloyed chromium-nickel steels [32] with the use of nickel and iron-based powders, the Vickers hardness is 460 MPa at the surface of the bed and decreases to 250 MPa at the base metal boundary (layer thickness is 0.5 mm).

TABLE V

Chemical composition (wt%)	remark	Ref.
C-0.2; Si-0.75; Cr-16; Ni-2.5; Mo-0.5; B-1.0; Co-0.5; Fe-bal	powder	[30]
Ti_6Al_4V -80; B_4C -20	powder	[31]
C-0.23; Mn-0.7; Si-0.7; Cr-13; Ni-2.2; Fe-bal	powder	[32]
C-0.34; Mn-0.6; Si-0.5; Cr-13; Ni-0.3; Fe-bal	powder	
Fe-3.7; C-0.5; Si-3.1; Cr-15; Ni-bal	powder	

Fe-4.9; C-0.7; Si-3.1; Cr-16; Ni-bal	powder	
Ti+TiC	powder	[33]
N	gas	[34]
C - graphite powder	paste	[35]
C; Ni=8C; V=4C; Ti=4C	paste	[36]
C; Ti	paste	[37]
Al; Ti; TiB ₂ ; TiC; SiC	paste	[39]

The deposition of layers by electroless method is realized in [40] ÷ [42]. The method consists of depositing a nickel layer followed by deposition of a layer containing a hardening phase - TiN or a detonating nano diamond. The particles of the hardening phase have nanoscale values of $4 \div 6 nm$. Low alloy steel and ductile cast iron are used as substrates. The hardness reached $720 HV_{0.3}$ and the wear resistance of the layers containing the hardening phase is up to 10 times higher than the nickel-only coating. Heat treatment ($290^{\circ}C / 6h$) can reduce wear to 36%.

Friction stir surfacing is also used for depositing layers on aluminium alloys [43] ÷ [45]. The additive material may be in the form of a wire or powder containing a hardening phase such as SiC or Al_2O_3 . Using a wire with a composition (wt.%): 97.5Al, 1.87Mg, 0.086Si, 0.006Mn, 0.009Cu, 0.27Fe and 0.21Cr a hardness of $85 HV_{0.2}$ [43] was achieved. When using a hardening phase, wear is reduced to 7 times compare to substrate.

In some of the publications reviewed, there is an indication of the coefficient of friction and its change in the process of the experiment [3] ÷ [5], [33], [35] and [44] ÷ [46]. These values varies from 0.12 to 0.75, and in some cases are averaged only.

VI. CONCLUSIONS

The technologies used to apply surface layers in the reviewed publications are electroslag surfacing, electric arc methods (MIG/ MAG, TIG, submerged arc and plasma surfacing), beam methods (electron beam and laser surfacing), friction stir electroless surfacing. Steel, aluminum and titanium alloys are used as substrates. The hardness (up to 61 HRC and 2300 HV) and wear resistance varies over a wide range depending on the process and the filler materials.

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